

## MALONYLATED ANTHOCYANINS IN *VERBENA* FLOWERS

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**Key Word Index**—*Verbena hybrida*; Verbenaceae; malonic acid; delphinidin 3-*O*-(6''-*O*-malonyl- $\beta$ -D-glucopyranoside); pelargonidin 3-*O*-(6''-*O*-malonyl- $\beta$ -D-glucopyranoside); flower colour.

**Abstract**—The major anthocyanin in the maroon flowers of *Verbena hybrida* cv. 'Splendor' was identified as delphinidin 3-*O*-(6''-*O*-malonyl- $\beta$ -D-glucopyranoside) by chromatographic and spectral methods. The major anthocyanin of scarlet cultivar 'Blaze' was determined to be pelargonidin 3-*O*-(6''-*O*-malonyl- $\beta$ -D-glucopyranoside).

### INTRODUCTION

Scott-Moncrieff and Sturgess (1940) first reported delphinidin 3-glucoside in the maroon cultivar of *Verbena hybrida*, and delphinidin 3,5-diglucoside in the purple form [1]. From genetical studies of flower colours on *Verbena hybrida*, Beale *et al.* suggested that acylated anthocyanins occurred in the flowers of this plant [2]. Later, Harborne confirmed the presence of pelargonidin, cyanidin and delphinidin in a number of cultivars of *Verbena hybrida* [3]. Moreover, he showed the occurrence of the zwitterionic anthocyanins in this plant during an anthocyanin survey by electrophoresis [4], but the structures of the anthocyanins were not determined.

We, therefore reinvestigated the anthocyanin components in *Verbena* cultivars. This paper deals with the structural elucidation of malonylated pelargonidin 3-glucoside and delphinidin 3-glucoside from two cultivars of *Verbena hybrida*.

### RESULTS AND DISCUSSION

The major anthocyanin **1** was extracted from the maroon flowers of *Verbena hybrida* cv. 'Splendor', and the major anthocyanin **2** from the scarlet flowers of cultivar 'Blaze'. Both **1** and **2** extracted were purified by repeated

prep. paper chromatography, and followed by Sephadex LH-20 gel column chromatography.

On acid hydrolysis, **1** and **2** gave delphinidin and pelargonidin, respectively, as the aglycones [cf-3]. In addition, TLC data of these hydrolysates in various solvents indicated that the sugar was glucose. UV-VIS spectral features of **1** and **2** were closely similar to those of delphinidin 3-glucoside (Dp 3-G) and pelargonidin 3-glucoside (Pg 3-G), respectively (Table 1), although, their  $R_f$  values of TLC were clearly different. On electrophoretic analysis, pigments **1** and **2** moved towards the anode in buffer pH 4.4, whereas Dp 3-G and Pg 3-G remained at the origin. Furthermore, when both pigments **1** and **2** were dissolved in the HCl-MeOH solvent, they were gradually converted into Dp 3-G and Pg 3-G, respectively. IR spectra of **1** and **2** showed characteristic carbonyl absorption bands (**1**;  $1710\text{ cm}^{-1}$  and **2**;  $1730\text{ cm}^{-1}$ , respectively), indicating that **1** and **2** are acylated with aliphatic dicarboxylic acids [5, 6]. On alkaline hydrolysis of these pigments gave only malonic acid.

The deacylated anthocyanins of **1** and **2** were identified as Dp 3-G and Pg 3-G, respectively (Table 1). Partial acid hydrolysis of **1** and **2** suggested that they were composed of anthocyanidin 3-glucoside and malonic acid at the molar ratio of 1:1. This information was confirmed by the

Table 1. Chromatographic and spectral properties of *Verbena hybrida* anthocyanins

Pigments	$R_f \times 100$ in				In 0.01% HCl-MeOH			
	BAW	BHW	1% HCl	AHW†	$\lambda_{\text{max}}^{\text{UV}}$	$\lambda_{\text{max}}^{\text{VIS}}$	$E_{320}/E_{\text{max}}^{\text{VIS}}$	$\text{AlCl}_3$ shift
<b>1</b> *	33	19	03	13	276	544	27	+
<b>2</b> *	63	48	10	33	272	512	13	—
Dp 3-G	25	10	01	06	278	542	26	+
Pg 3-G	49	43	05	27	272	512	13	—

\* Pigment **1**, **2** = 3-*O*-(6''-*O*-malonyl- $\beta$ -D-glucopyranoside) of delphinidin and pelargonidin respectively; Dp 3-G, Pg 3-G = 3-glucoside of delphinidin and pelargonidin.

† TLC on microcrystalline cellulose; BAW = *n*-BuOH-HOAc-H<sub>2</sub>O (4:1:5), BHW = *n*-BuOH-2M HCl (1:1), AHW = HOAc-HCl-H<sub>2</sub>O (15:3:82).

**FAB-MS.** The FAB-MS of each anthocyanin gave the molecular ion  $[M]^+$  (as flavylium ion) peak at  $m/z$  551 or 519 with fragment of  $[M-86]^+$  and aglycone ion, indicating **1** and **2** to be monomalonylated anthocyanins.

In order to confirm these structures, the 400 MHz proton FT-NMR spectra were measured in 10%  $CF_3COOD$  and 90%  $DMSO-d_6$ . The NMR signals of both pigments clearly indicated the presence of malonyl  $-CH_2-$  protons at near  $\delta$  3.4 ppm; **1** at  $\delta$  3.37 ( $d$ ,  $J = 17$  Hz) and 3.40 ( $d$ ,  $J = 17$  Hz), and **2** at  $\delta$  3.35 ( $d$ ,  $J = 16$  Hz) and 3.40 ( $d$ ,  $J = 16$  Hz) [6]. For the aglycone moiety, the large differences of shifts were observed at H-2',  $\delta$  8.59 (or  $\delta$  8.61 [7]) for Pg, 8.04 for Cy [6] and 7.72 for Dp, indicating a consistent effect due to increasing hydroxylation on the B-ring. The signal patterns of sugar protons of both pigments were similar [7]. The characteristic two downfield shift protons at  $\delta$  4.13, 4.46 ppm in **1** and 4.12, 4.46 in **2** with the geminal coupling ( $J = 12$  Hz) were assigned to glucose methylene protons, and this fact indicated that the malonyl residue in each pigment was attached to C-6'' of glucose moiety [5–7]. Furthermore, each proton resonated at the lower magnetic field with large coupling constant (**1**;  $J = 8$  Hz and **2**;  $J = 8$  Hz), and the sugar protons had also large  $J$ -values such about 9 Hz. Therefore the glucose was proved to have a  $\beta$ -D-pyranose form in each pigment.

Hence, the major anthocyanin **1** in the maroon flowers of the cultivar 'Splendor' is delphinidin 3-O-(6''-O-malonyl- $\beta$ -D-glucopyranoside), which is a new malonylated pigment. The major pigment **2** in the scarlet 'Blaze' is pelargonidin 3-O-(6''-O-malonyl- $\beta$ -D-glucopyranoside), which has been already identified by Takeda *et al.* [8]. However, it is the first time that the 'Blaze' pigment structure has been fully determined by FAB-MS and NMR.

#### EXPERIMENTAL

**Plant material.** Flower petals of *Verbena hybrida* cv. Splendor and Blaze were collected at the farm of Minami-Kyushu University and dried at 50° overnight.

**Isolation and purification.** Dried petals (ca 100 g) of each cultivar were extracted with  $EtOH-HOAc-H_2O$  (10:1:10), and the extract concd to a small vol. at 40°. The concentrate was purified by prep. PC (Toyo No. 526 filter paper) in the solvents 15%  $HOAc$  and  $n-BuOH-HOAc-H_2O$  (6:1:2), and followed by Sephadex LH-20 CC with 15%  $HOAc$  in order to separate flavonols and other phenolic components. Then, pigment **1** (ca 40 mg) and **2** (ca 60 mg) were obtained respectively.

**Standard analysis.** Characterization of pigment **1** and **2** was carried out by Avicel cellulose TLC, paper electrophoresis and UV-VIS spectrometry [4, 8]. Moreover, these pigments were analysed by TLC after the processes of acid hydrolysis, alkaline deacylation, and partial acid hydrolysis.

**IR and FAB-MS.** IR spectra were measured as the KBr discs. Positive FAB-MS spectra were recorded with glycerol as a matrix.

**Delphinidin 3-O-(6''-O-malonyl- $\beta$ -D-glucopyranoside) (1).** IR  $\nu_{max}^{KBr} cm^{-1}$ : 3400, 1710 ( $C=O$ ), 1625, 1380, 1060; FAB-MS  $m/z$  (rel. int.): 551  $[M]^+$  (24) ( $C_{24}H_{23}O_{15}$  requires 551, as flavylium ion), 465  $[M-86]^+$  (4) (loss of malonyl group), 303 [delphinidin] $^+$  (11);  $^1H$  NMR (400 MHz, 10%  $CF_3COOD$  in  $DMSO-d_6$ ):  $\delta$  aglycone moiety: 8.77 (1H, s, H-4), 7.72 (2H, s, H-2' and 6'), 6.86 (1H,  $d$ ,  $J = 2$  Hz, H-8), 6.72 (1H,  $d$ ,  $J = 1$  Hz, H-6), glucose moiety: 5.44 (1H,  $d$ ,  $J = 8$  Hz, H-1''), 4.46 (1H,  $dd$ ,  $J = 12, 2$  Hz, H-6''a), 4.13 (1H,  $dd$ ,  $J = 12, 8$  Hz, H-6''b), 3.86 (1H,  $ddd$ ,  $J = 9, 8, 2$  Hz, H-5''), 3.60 (1H,  $dd$ ,  $J = 9, 8$  Hz, H-2''), 3.43 (1H,  $t$ ,  $J = 9$  Hz, H-3''), 3.26 (1H,  $t$ ,  $J = 9$  Hz, H-4''), malonyl  $-CH_2-$  moiety: 3.40 (1H,  $d$ ,  $J = 17$  Hz), 3.37 (1H,  $d$ ,  $J = 17$  Hz).

**Pelargonidin 3-O-(6''-O-malonyl- $\beta$ -D-glucopyranoside) (2).** IR  $\nu_{max}^{KBr} cm^{-1}$ : 3400, 1730 ( $C=O$ ), 1635, 1330, 1170, 1060; FAB-MS  $m/z$  (rel. int.): 519  $[M]^+$  (66) ( $C_{24}H_{23}O_{13}$  requires 519, as flavylium ion), 433  $[M-86]^+$  (5) (loss of malonyl group), 271 [pelargonidin] $^+$  (20);  $^1H$  NMR (400 MHz, 10%  $CF_3COOD$  in  $DMSO-d_6$ ):  $\delta$  aglycone moiety: 8.88 (1H, s, H-4), 8.59 (2H,  $d$ ,  $J = 9$  Hz, H-2' and 6'), 7.07 (2H,  $d$ ,  $J = 9$  Hz, H-3' and 5'), 6.99 (1H,  $d$ ,  $J = 1$  Hz, H-8), 6.74 (1H,  $d$ ,  $J = 2$  Hz, H-6), glucose moiety: 5.40 (1H,  $d$ ,  $J = 8$  Hz, H-1''), 4.46 (1H,  $dd$ ,  $J = 12, 2$  Hz, H-6''a), 4.12 (1H,  $dd$ ,  $J = 12, 8$  Hz, H-6''b), 3.82 (1H,  $ddd$ ,  $J = 9, 8, 2$  Hz, H-5''), 3.49 (1H,  $dd$ ,  $J = 9, 8$  Hz, H-2''), 3.42 (1H,  $t$ , 9 Hz, H-3''), 3.24 (1H,  $t$ ,  $J = 9$  Hz, H-4''), malonyl  $-CH_2-$  moiety: 3.40 (1H,  $d$ ,  $J = 16$  Hz), 3.35 (1H,  $d$ ,  $J = 16$  Hz).

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